
OPTIMIZING OF NANO-FILTRATION TO OBTAIN FISH PROTEIN ISOLATE (FPI) FROM *PANGASIU*S *HYPOPHTHALMUS* BYPRODUCTS WITH CALCIUM-BINDING BIO-ACTIVITY**¹C. X. Thuy, ²T. B. Lam, ³K. Mc. Commick**¹ Faculty of Food Technology, Hochiminh City University of Food Industry (HUFI)- Vietnam² Department of Food Technology, HCMC University of Technology (VNU) - Vietnam³ Minnesota University - USA

ABSTRACT: *The by-products from Pangasius hypophthalmus in fishery processing plants in the Mekong River Delta, Vietnam is very abundant but not being fully utilized. Calcium-binding is one of the most important biological functions of FPI which derived from Pangasius hypophthalmus byproducts. This opens up prospects for using of FPI from catfish byproducts to manufacture the products enhancing human calcium absorption. This study focused on optimization of experimental factors during nano-filtration (NF). The separation has proceeded by using 5 kDa GE-5-DL spiral NF membrane at difference operated conditions and optimized using response surface methodology (RSM) to obtain FPI from Pangasius hypophthalmus by-products with the highest possibility of calcium-binding. Results showed that the highest calcium-binding of FPI is 31.54 (mg Ca⁺²/g FPI) with the NF conditions: temperature 45°C, input flow's velocity 29 l/h; operating pressure 11 bar. The protein percentage by molecular weight of FPI as follows: <3 kDa: 82.88%, from 3 kDa to <7 kDa: 17.12 %. Type the links between Ca⁺² and proteins in FPI: 93.52% binding through EF hand structure; 1.18% as hydrogen bonding through water bridge; 3.24% as electrostatic links between Ca⁺² with the starting amino acid of proteins.*

KEYWORDS: GE-5-DL, FPI, Calcium-Binding, Pangasius Hypophthalmus By-Products, Optimization (RSM).

INTRODUCTION

FPI contains proteins with small molecular weight. It can be called peptone from fish (Asbjorn Gildberg, 2004). FPI's functional properties as well as biological activities depend on its origin and produced methods (Samanta S. Khora, 2013). Several studies have shown that FPI's proteins derived from *Siniperca scherzeri*, surimi of *Hwangtae* (*yellowish dlied pollack*) after binding with calcium in the form of ions, are added to food to enhance the calcium absorption of users (San Soon chol et al., 2008; Wuying Chu et al., 2013). Calcium-binding possibility of FPI from *Mallotus villosus* by-products is equivalent to the one of whey protein isolate - WPI (Hannu Korhonen et al., 2006). The ability of FPI's calcium-binding is determined by two main factors: the structure

of proteins (the number, composition, sequencing arrangement of amino acids) and the interaction among the proteins in FPI (David M. Balshaw et al., 2001; Rong Liu et al., 2011). FPI's peptides with a length of about 7 to 30 amino acids would have the best calcium-binding ability commonly (Laurent Picot et al., 2010; Hoa M.X., 2012). The Proteins and interaction among themselves could form EF hand structure. At EF hand structural position, calcium ions can be binded to proteins. The Ca^{+2} ions have been binded to proteins by ionic bonds, static links, or hydrogen bonds through the intermediate bridge of water etc... (M. Susan Cates et al., 2002; Tai M.V., 2013; Ruiyan Nie et al., 2014)

Currently, the *Pangasius hypophthalmus* by-products from fishery processing plants in the Mekong River Delta, Vietnam is enormous (over 1 million tons/year) (VASEP, 2014). These by-products have been mainly used as animal feed (Hoa M.X., 2013). The study how to turn the *Pangasius hypophthalmus* by-products to value-added products, such as FPI with high calcium-binding ability will bring significant economic efficiency for manufacturers and farmers.

METHODOLOGY

METERIAL

Hydrolysis solution and FPI

Hydrolyzing the *Pangasius hypophthalmus* by-products by protease (Alcalaze 2.4L) at optimum conditions (pH: 7.0; Enzyme/Substrate ratio (E/S): 0.15 % w/v ; hydrolysis temperature: 55⁰C; hydrolysis time: 120 minutes). Hydrolyzed solution was then cooled to 4⁰C for a preliminary de-fatting, vacuum filtered through non-ash paper and then centrifuged to de-fat at the speed of 15,000 rpm for 20 minutes. The solution obtained after centrifugation was brought to filter by NF membrane. The permeat was freeze-dried to get FPI powder.

Determination of molecular weight of the proteins in FPI were by high pressure liquid chromatography (HPLC)

NF membrane

Using 5 kDa GE-5-DL spirral membrane (imported from GE Osmonics, USA). Total filtering area is 0.325 m²; maximum operating temperature is 50⁰C; size (diameter x length) is 30.32x457.2 cm, maximum input flow is 6.53 l/min; maximum operating pressure is 41 bar.

MF membrane (for preliminary filtering)

Using HP3500 membranes (imported from Stirlitech, USA) Total surface area is 0.1556 m², maximum operating temperature is 50⁰C.

Protease and chemicals

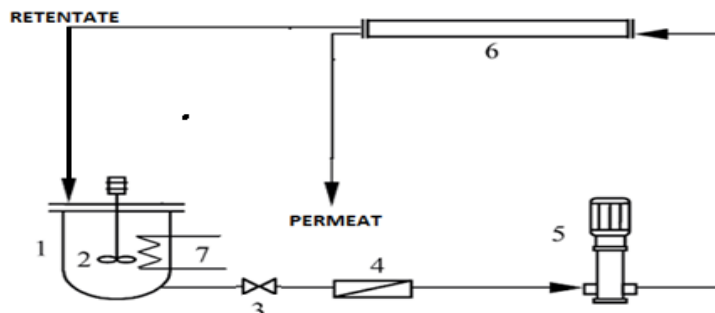
Enzymes Alcalase 2.4L were purchased from EAC Co., Ltd. (sole-exclusive agent for Novozyme in Ho Chi Minh city, Vietnam).

All chemicals reagents used for the experiments were in analytical grade

METHODS

Arrangement of membrane systems

Membrane schematics:



Where, (1) input tank, (2) stirrer, (3) controlled-valve, (4) microfiltration (MF) membrane (preliminary filter), (5) high-pressure pump; (6) nanofiltration (NF) membrane; (7) the instrument for adjusting the input solution temperature.

Membrane operation

Input flow was heated to proper temperature by adjusting instrument (7), preliminarily filtered by MF membrane (4), then solution will be pumped to the NF membrane (6) by high-pressure pump (5). The retentate was refluxed 10 times without diafiltration technology. The permeate (including proteins with molecular weight are smaller than 5 kDa) will be freeze-dried for collecting FPI powder. FPI powder was used for studying the calcium-binding ability as well as protein's molecular weight.

Determination of FPI's calcium-binding possibility

Calcium-binding possibility of FPI is determined by the method of Flame Atomic Absorption Spectrophotometric (FAAS).

For binding Ca^{+2} ions to proteins in FPI

1g FPI powder was dissolved in 1 liter of sodium phosphate buffer solution (pH = 7.8); added 1.11 g CaCl_2 (20 μ) to FPI solution; adjusted the temperature to about 20 - 22 $^{\circ}\text{C}$; stirred for 30 minutes at speed of 100 rpm; preliminarily centrifuged at 6000-7000 rpm, adjust pH = 7 with bicarbonate buffer solution; super-speed re-centrifuged at 26,000 rpm.

Calculation the calcium-binding possibility of FPI (mg Ca^{+2} /g FPI)

Vaporizing and atomization samples by gas flame. The gas clouds will absorb the monochromator radiation beam. Using the spectrometer for collecting the entire separation spectrum beams and select a calcium spectral absorption line in order to measure its intensity. In a certain limit of concentration (C) of substances to be determined. The value of this intensity depends linearly on the concentration (C) according to the following equation:

$$A = a.C; \text{Asb} = \lg = \text{ep.l.c I0/It}$$

Where: A is the intensity of the spectral absorption; a is experimental constant; C is concentration of substances to be determined (mg/g)

Testing the links between Ca^{+2} and proteins in FPI

- After Ca^{+2} associated with proteins in FPI (section 2.2.2.), to ensure that Ca^{+2} were binded to proteins by ionic bonds, static links, or hydrogen bonds through the intermediate bridge of water

etc... (Not a mechanical bond). The solution obtained after ultracentrifugation at 26000 rpm will added 25ml LYSIS buffer solution (Consists of 50 mμ Tris-H₃PO₄; 2 mμ EDTA, 1 mμ phenyl methyl sulfonyl fluoride (PMSF), 1microgam leupeptin).

- The solution was put into a magnetic field; cooled at 4⁰C for 5 minutes to ensure the complete removal of CaCl₂ that were stick or not binded to a protein in FPI; Adjusted the pH to 7.5; brought the solution to monochromatic radiation environment. The spectral lines of calcium ions will be magnified 20000 times to know the density of calcium-binding to the FPI's proteins.

Optimization of NF operation to FPI's calcium-binding ability

Experimental planning: Response Surface Methodology (RSM) with 2 fold rotation plan centered (star distance $\alpha = 1.682$) is applied to optimize the conditions of NF operation. In particular, preliminary experimental has previously been selected as the basis for experimental optimization design. The parameters include: temperature, velocity and pressure of input flow.

RESULTS AND DISCUSSION

Optimization of NF operation

Before optimizing, we conducted preliminary experiments to identify the influence of each individual factor (temperature, velocity and pressure of input flow) to FPI's calcium-binding ability. The results of preliminary experiments were used as the basis for determining the simultaneously impact by all of three above factors to FPI's calcium-binding ability

When designing the optimization modal, the temperature of input flow (X_1 , ⁰C), velocity of input flow (X_2 , litter/h), and pressure of input flow (X_3 , bar) were experimented simultaneously for 2 purposes: (1) to building the regression equation that describe the relationship among the NF operation's factors that influence to the FPI's calcium-binding ability from *Pangasius hypophthalmus* by-products. (2) Finding out the NF operation's conditions in order to achieve the maximum FPI's calcium-binding ability. The experiment levels of independent factors in optimization as follows (Table 1):

Table 1. Levels of independent factors in optimization

Factors	Levels of factors				
	- α	-1	0	+1	+ α
Temperature of input flow (X_1 , ⁰ C)	37	40	45	50	53
Velocity of input flow (X_2 , l/h)	17	20	25	30	33
Pressure of input flow (X_3 , bar)	7	8	10	12	13

The coded levels of variables: above level (+1); database level (0); below level (-1); $\alpha = 1,682$

The experimental number were calculated as follows: $N = 2^k + 2k + n_0 = 2^3 + 2.3+6 = 20$ (k: number of experimental factors (k = 3), n_0 : number of experiments in center or mind ($n_0 = 6$). Experimental matrix shows the simultaneously of the factors is presented in Table 2.

Table 2. Experimental matrix

No	Coded levels of variables			Real variables			Observed calcium-binding ability (mg/g FPI)	Expected calcium-binding ability (mg/g FPI)
	x1	x2	x3	X1	X2	X3		
1	-1	-1	-1	40	20	8	22.61	22.59
2	1	-1	-1	50	20	8	25.22	25.35
3	-1	1	-1	40	30	8	25.87	26.15
4	1	1	-1	50	30	8	28.81	28.43
5	-1	-1	1	40	20	12	25.22	25.77
6	1	-1	1	50	20	12	25.87	25.76
7	-1	1	1	40	30	12	29.13	29.17
8	1	1	1	50	30	12	28.48	28.67
9	-1.682	0	0	36.59	25	10	24.57	24.15
10	1.682	0	0	53.41	25	10	25.87	26.05
11	0	-1.682	0	45	16.59	10	25.22	24.98
12	0	1.682	0	45	33.41	10	30.44	30.44
13	0	0	-1.682	45	25	6.64	25.87	25.94
14	0	0	1.682	45	25	13.37	29.13	28.82
15	0	0	0	45	25	10	30.96	30.80
16	0	0	0	45	25	10	31.09	30.80
17	0	0	0	45	25	10	30.76	30.80
18	0	0	0	45	25	10	30.41	30.80
19	0	0	0	45	25	10	30.44	30.80
20	0	0	0	45	25	10	31.09	30.80

Regression equation is a second-order polynomial:

$$Y = \beta_0 + \sum_{i=1}^3 \beta_i X_i + \sum_{i=1}^3 \beta_{ii} X_i^2 + \sum_{i=1}^3 \sum_{j=i+1}^3 \beta_{ij} X_i X_j$$

Where, Y is the dependent variable (calcium-binding ability in real value); X_i and X_j are the levels of the independent variable (experimental factor) which represent the influence of X_1 , X_2 , X_3 on the response factor (calcium-binding ability); β_0 is constant; β_i , β_{ii} , β_{ij} are the coefficients of the regression equation.

To build the mathematical description as a regression equation, the coefficients of the equation must be determined. Its coefficients have the following values:

$$b_0 = 30.80 \quad b_1 = 0.57 \quad b_2 = 1.62 \quad b_3 = 0.85 \quad b_{11} = -2.01 \quad b_{22} = -1.09$$

$$b_{33} = -1.21 \quad b_{12} = -0.12 \quad b_{13} = -0.69 \quad b_{23} = 0.04$$

Two of coefficients: b_{12} ($P=0.76459 > 0.05$) and b_{23} ($P=0.386967 > 0.05$) have no statistical significance. Thus, these coefficients (b_{12} and b_{23}) are removed from the regression equation. The regression equation takes the following form:

$$Y = 30.80 + 0.57 X_1 + 1.62 X_2 + 0.85 X_3 - 2.01 X_1^2 - 1.21 X_2^2 + 0.47 X_3^2 - 0.69 X_1 X_3$$

Testing the compatibility of the regression equation and experimental results shows that three experimental factors (X_1 , X_2 , X_3) have a strong influence ($P < 0.05$) on the calcium-binding ability of FPI from *Pangasius hypophthalmus* by-products during NF operation. The compatibility of the

regression equation (Lack of fit) is checked with the Modde 5.0. After checking, the "Lack of fit" is not statistically significant (Lack of fit has $P = 0.23$, $P > 0.05$). Thus the regression equation has high compatibility with experiments (Kun-Nan Chen, 2008)

Table 3. Testing results the compatibility of the regression equation

Source	Degree of Freedom (DF)	Sum of Squares (SS)	Mean Squares (MS)	F value	P value	
Regression	9	136.91	15.2123	105.833	0.000	significant
Residual	10	1.43738	0.143738			
Lack of Fit	5	0.960699	0.19214	2.01539	0.23	insignificant
Pure Error	5	0.476682	0.0953365			
Total Corrected	19	138.348	7.28146			

The influence of the experimental factors on the calcium-binding ability of FPI from *Pangasius hypophthalmus* by-products are shown on the contour and response surface in Figure 1

Overall, the calcium-binding ability of FPI increased gradually along with the rising of experimental factors to a limited value during NF operation. However, then the calcium-binding ability was stability and it is tended to decrease when the values of factors exceeded the limit value. Figure 1-B shows the effects of pair of elements (X_1 : temperature of input flow and X_3 : pressure of input flow) on FPI's calcium-binding ability. Results indicate that calcium-binding ability rises up about 25% plus, along with the increasing of the temperature (to 45°C) and pressure (to 11 bar). However, the calcium-binding ability tended to reach a stable value and then reduced lightly when NF operation was carried out in condition of higher temperature and pressure.

Figure 1-A showed the influence of the remaining pairs of elements: temperature (X_1) and input flow's velocity (X_2) to calcium-binding ability of FPI. At temperatures between 40°C and 45°C, the calcium-binding ability increased nearly 19% (from 25.38 mg Ca^{+2} /g FPI to 31.38 mg Ca^{+2} /g FPI). However, when input flow's velocity continued to rise, calcium-binding ability decreased lightly. The same rule is observed (in Figure 1-C) when considering simultaneously the impact of X_1 and X_2 (velocity and pressure of input flow respectively).

This is explained as follows: as the increasing of temperature, the movement speed of the proteins in solution also increased. Pressure and velocity of input increased making the rising of both flux through the membrane and the separation level of proteins in FPI (Laurent Vandanjon et al., 2007; Anusha Geethangani Perera Samaranayaka, 2010). So the number of proteins with small molecular weight in permeate would be increased. The group of small molecular weight proteins (1-3 kDa) has the highest calcium-binding ability (Hoa M.X., 2012; P. Bourseau et al., 2009). This causes the rising of FPI's calcium-binding possibility. However, when we increased the input flow's velocity and pressure without application of diafiltration technology would cause the fouling and concentration. The small molecular weight proteins in permeate tended to be stable. So the FPI's calcium-binding will hardly increase. On the other hand, prolonging NF operation time can affect the interaction of the proteins in solution (Tai M.V., 2013). This influence should decrease calcium-binding of FPI.

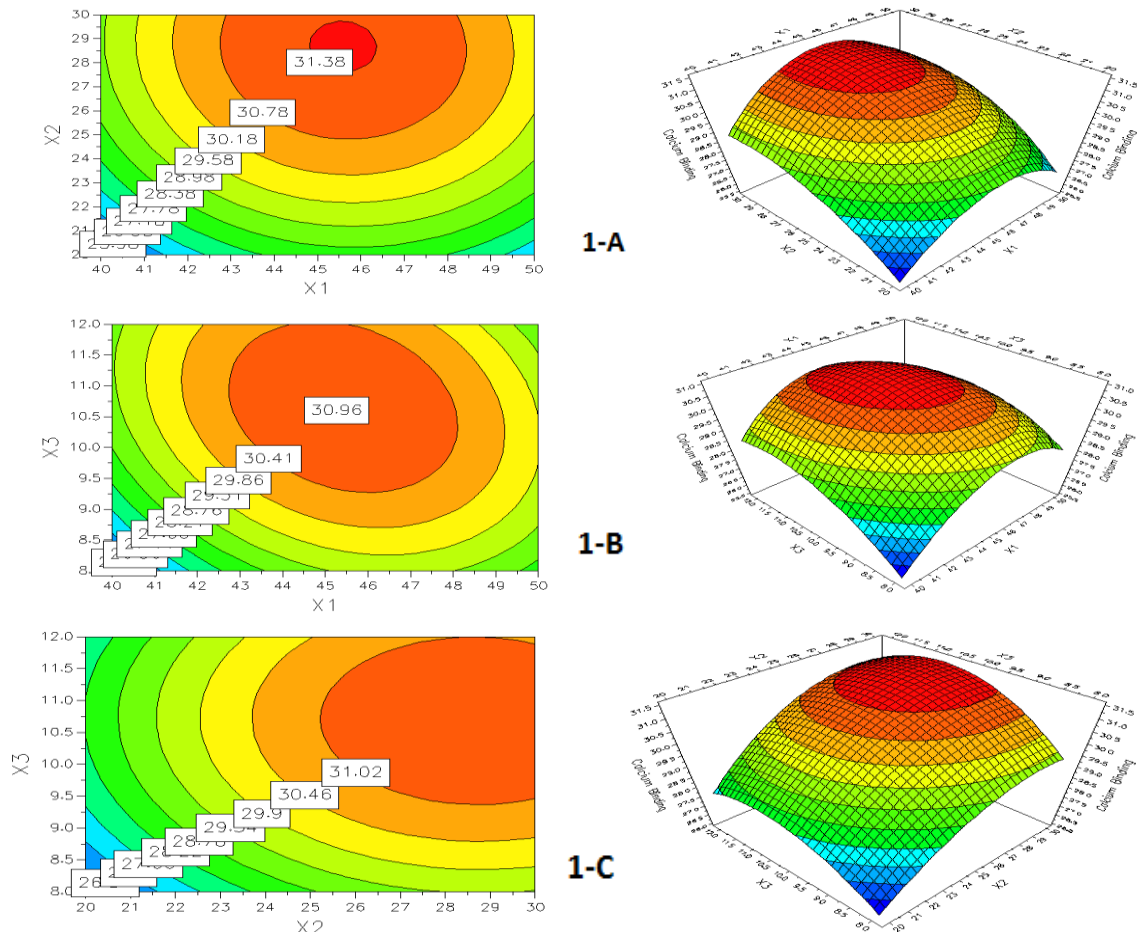


Figure 1. Contour and Response Surface showing the influence of three experimental factors on the FPI's calcium-binding ability.

Optimization results show that the calcium-binding ability of FPI from *Pangasius hypophthalmus* by-products reaches the highest of 31,54 (mg Ca^{+2} /g FPI) in terms of NF operation factors: temperature is 45⁰C, velocity of input flow is 29 l/h, input flow's pressure is 11 bar.

The above optimization results are similar to previous studies have been published for calcium-binding of FPI from *Sardinella aurita*, *zebrafish* and *rainbow trout*. The calcium-binding ability of FPI from *Silver carp* is 31.09 calcium (Ca^{+2} mg/g FPI) with the NF operation factors: temperature of input flow 45⁰C, pressure of input flow 12 bar, velocity of input flow 25 l/h (Tai M.V., 2013; Zhengjin Cao et al., 2003). Calcium-binding ability of FPI from *Salmon's* surimi is 29.93 (mg Ca^{+2} /g FPI) in the operating NF conditions: temperature of input flow 42⁰C, pressure of input flow 10 bar, velocity of input flow 25 l/h (Laurent Picot et al., 2010). Our findings about Calcium-binding ability of FPI from *Pangasius hypophthalmus* by-products has been similar to the studying results about the calcium-binding ability of FPI that derived from *Pangasiidae* by-products, *blue whiting* surimi. This one reached 30.02 mg Ca^{+2} /g FPI in NF operating conditions: temperature of input flow 45⁰C, pressure of input flow 11 bar, velocity of input flow 28 l/h (Hoa M. X., 2012; P. Laurent Vandanjon et al., 2007)

The analytical results of the molecular weight of proteins in FPI from *Pangasius hypophthalmus* by-products

Analytical results of the molecular weight of proteins in FPI that derived from *Pangasius hypophthalmus* by-products in the NF operating conditions after optimization (temperature: 45°C, velocity of input flow: 29 l/h, input flow's pressure: 11 bar) to have the highest calcium-binding ability 31,54 (mg Ca²⁺/g FPI) are shown in table 4 and Figure 2.

Table 4. The ratio distribution basing on molecular weight of proteins in FPI

No	Molecular weight (kDa)	Ratio (%)
1	> 20	0
2	10 ÷ 20	0
3	7 ÷ 10	0
4	3 ÷ <7	17.12
5	< 3	82.88

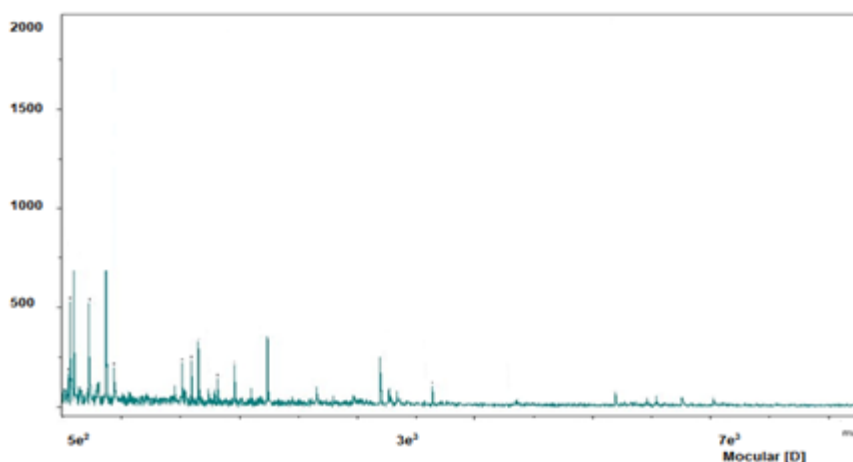


Figure 2. HPLC analysis results determining the molecular weight distribution of proteins in FPI

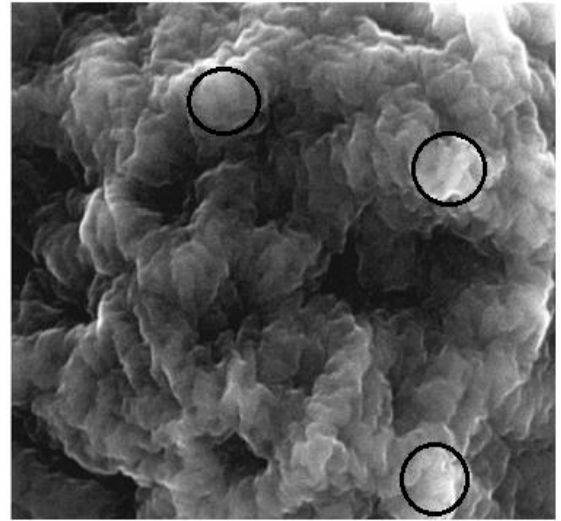
According to results in Table 4, when the input flow's pressure reached 11 bar, the proteins in FPI in the molecular weight of 7 kDa or smaller than accounted for 17.12%, in which the proteins that molecular weight less than 3 kDa are dominant (82.88 %). This group of peptide plays a decisive role to calcium-binding ability of FPI. The rate of proteins in molecular weight <3 kDa accounted for approximately 60 % of the FPI's total protein would have an important role in improving the calcium-binding ability of FPI. This result is similar to studyings of Tai M.V., 2013; Laurent Picot at al., 2010. They confirmed that calcium-binding ability of FPI obtained from *Sardinella aurita* or *Clupea harengus* or *Gasdus morhua* reached the highest value when quantity of protein with molecular weight from 2 kDa to 8 kDa around 70%.

The links between Ca²⁺ and proteins in FPI from *Pangasius hypophthalmus* by-products

Results of testing the links between Ca²⁺ with FPI's proteins are shown in Table 5 and Figure 3. Table 5 shows that in condition of optimal values of factors during the NF separation, especially when velocity and pressure of input flow reached 29 l/h and 11 bar respectively, the number of small molecular weight proteins in permeat has gained the maximum value. The small molecular weight proteins (mainly from 1 to 3 kDa) play a key role in FPI's calcium-binding ability. The small-sized proteins and interaction among themselves may create many EF hand. Each EF hand can combine with Ca²⁺ ions by covalent bonds, ionic bonds, hydrogen bonds through the "bridge" of water or electrostatic links.

Table 5. Results of testing the links between Ca²⁺ and proteins in FPI

Area [%]	Ca ²⁺ – Binding to protein	Comment
93.52	Binding constants of Ca ²⁺ Ion to Calcium-binding protein in high density	<i>EF-hand effects predicted</i>
1.18	Binding constants of Ca ²⁺ Ion to Calcium-binding protein in medium density	<i>Hydrogen bonding via H₂O</i>
3.24	Binding constants of Ca ²⁺ Ion to Calcium-binding protein in low density	<i>Peptide started by no loop, -C-C-structure</i>
2.06	not identified	

**Figure 3. The links between Ca²⁺ and FPI's proteins**

At the optimal values of the NF operation, there were 93.52 % Ca²⁺ in solution binded to proteins via forming of EF hand structure; 1.18 % linkages between Ca²⁺ with the proteins were hydrogen bonds through the “bridge” of water; 3.24 % links among Ca²⁺ and proteins are associated by electrostatic bonds (between Ca²⁺ with the proteins that started by no loop amino acid and beginning – C–C – structure (such as: Leu , Met , Arg...)). This result is similar to findings of Tai M.V., 2013; Ann Elizabeth Theodore, (2005); Rong Liu et al., 2011. The authors confirmed that the calcium-binding ability of FPI derived from freshwater fish *Blunt Snout Bream (Megalobrama amblycephala)*, by-product of *Sardinella aurita* or surimi of some *catfish*, mostly formed through the structure of the EF hand (over 79 %); about 6 % Ca²⁺ binded to proteins via electrostatic bonds between Ca²⁺ and amino acids in protein's structure .

CONCLUSION

FPI derived from *Pangasius hypophthalmus* by-products through NF operation has good calcium-binding ability. The calcium-binding ability achieves maximum value 31,54 mg Ca²⁺/g FPI, corresponding to the NF operation conditions as follows: input flow's temperature is 45⁰C, velocity of flow input is 29 l/h, input flow's pressure is 11 bar. Type of binding between mg Ca²⁺ and FPI's proteins: 93.52% via EF hand structures; 1:18% as hydrogen bonding through the medium-bridge of water; 3.24% as electrostatic links between Ca²⁺ and the proteins that started by amino acid with – C–C – and no loop structure. The protein percentage by molecular weight of FPI as follows: <3 kDa: 82.88%, from 3 kDa to <7 kDa: 17.12 %.

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